This article was downloaded by:

On: 26 January 2011

Access details: Access Details: Free Access

Publisher *Taylor & Francis*

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-

41 Mortimer Street, London W1T 3JH, UK



Nucleosides, Nucleotides and Nucleic Acids

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713597286

Oligonucleotides and Nucleotide-Peptides. LIV. The Synthesis and Hydrolytic Properties of Adenylyl-(5' \rightarrow N $_{\epsilon}$)-lysylpeptides — Model Compounds of the AMP-RNA Ligase Covalent Complex

Benediktas Juodka^a; Sofija Sasnauskiene^a; Ruta Petniunaite^a

^a Department of Biochemistry and Biophysics, Vilnius University Vilnius, Lithuania

To cite this Article Juodka, Benediktas , Sasnauskiene, Sofija and Petniunaite, Ruta(1992) 'Oligonucleotides and Nucleotide-Peptides. LIV. The Synthesis and Hydrolytic Properties of Adenylyl-(5' \rightarrow N)-lysylpeptides — Model Compounds of the AMP-RNA Ligase Covalent Complex', Nucleosides, Nucleotides and Nucleic Acids, 11: 8, 1445 — 1453

To link to this Article: DOI: 10.1080/07328319208021186 URL: http://dx.doi.org/10.1080/07328319208021186

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

OLIGONUCLEOTIDES AND NUCLEOTIDE-PEPTIDES. LIV. THE SYNTHESIS AND HYDROLYTIC PROPERTIES OF ADENYLYL-(5'→N_e)-LYSYLPEPTIDES -- MODEL COMPOUNDS OF THE AMP-RNA LIGASE COVALENT COMPLEX¹

Benediktas Juodka*, Sofija Sasnauskiene, Ruta Petniunaite Department of Biochemistry and Biophysics, Vilnius University Vilnius, Ciurlionio 21, Lithuania, 232009

Abstract: RNA ligase active center model compounds - adenylyl- $(5\rightarrow N_e)$ -lysylpeptides were synthesized. The stability of these compounds in aqueous solutions was studied and it was shown, that the carboxyl group of lysine and glutamic acid or hydroxyl group of threonine have no effect upon the hydrolytic mechanism of the adenylyl- $(5\rightarrow N_e)$ -lysylpeptides. This led to conclusion, that the hydrolysis of the AMP-RNA ligase complex is dependant upon other amino acid functional groups, which may be located next to the phosphoamide center, as a result of tertiary protein structure.

RNA ligase from T4 phage infected *Escherichia coli* (polinucleotidyl synthetase, EC 6.5.1.3) was first described in 1972.² The enzyme has been found to catalyze the ATP-dependant circularization and intermolecular joining of oligonucleotides.^{3,4} This catalysis is found to occur in three stages:

1. E + ATP
$$\leq \frac{Mg^{2+}}{2}$$
 E-pA + PP_i

2.
$$E-pA + pN(pN)_n \leq [A(5')ppN(pN)_n] \cdot E$$

3.
$$N(pN)_m pN + [A(5')ppN(pN)_n] \cdot E \longrightarrow N(pN)_m pNpN(pN)_n + AMP + E$$

In the first stage of the reaction the RNA ligase reacts with ATP to form the covalent intermediate — AMP-RNA ligase complex (E-pA) with the liberation of free pyrophosphate.⁵ Further data provided evidence for the formation of the phosphoamide bond between AMP and

¹<u>Abbreviations</u> used for amino acids and peptides follow the rules of the IUPAC-IUB Commission of Biochemical Nomenclature <u>J. Biol. Chem.</u> 1972 247, 977. The following additional abbreviations are used: Boc, tret-Butoxycarbonyl; Cbz, Benzyloxycarbonyl; DMF, N,N'-dimethylformamide; DMSO, dimethylsulphoxide; TEA, triethylamine; DCC, N,N'-dicyclohexylcarbodiimide.

RNA ligase via lysine ϵ -amino group.⁶ The covalent complex (E-pA) has also been found to undergo hydrolysis in aqueous solution, where adenosine and AMP are identified as decomposition products.⁷

Studies conducted with model phosphoamide type nucleotidyl-(P-N)-amino acids or nucleotidyl-(P-N)-peptides showed that phosphoester bond cleavage (adenosine elimination) occurs only under the conditions where the amino acid residue hydroxyl or carboxyl group is located next to the phosphoamide center. Based upon the amino acid sequence of RNR ligase, 9,10 it has been shown that lysine-99 (to which AMP is bound) is found between threonine-98 and glutamic acid-100. The phosphoamide center of the AMP-RNA ligase covalent complex is localized between two potential functional groups which may participate in intramolecular nucleophilic catalysis. It is therefore possible that their existence is responsible for the unexpected cleavage which the phosphoamide center undergoes during the hydrolysis of the AMP-RNA ligase complex.

The purpose of this work has been to synthesize adenylyl- $(5' \rightarrow N_{\epsilon})$ -lysine and adenylyl- $(5' \rightarrow N_{\epsilon})$ -lysylpeptides, which would serve as model compounds for the AMP-RNA ligase covalent complex. Furthermore, the influence of the threonine hydroxyl and lysine or glutamic acid carboxyl groups upon the stability of the phosphoamide center was studied.

RESULTS AND DISCUSSION

The protected dipeptides Boc-Lys(Cbz)-Glu(OEt)-OEt (I) and Boc-Thr-Lys(Cbz)-OMe (II) were synthesized according to the N,N'-dicyclohexylcarbodiimide (DCC) method in solution:

Boc-NH-CH-COOH + NH₂-CH-COOC₂H₅
$$\xrightarrow{DCC}$$

(CH₂)₄ (CH₂)₂

NH-Cbz COOC₂H₅

Boc-NH-CH-CO-NH-CH-COOC₂H₅

(CH₂)₄ (CH₂)₂

NH-Cbz COOC₂H₅

II. Boc-Thr-Lys(Cbz)-OMe

The structure of the dipeptides (I, II) was determined by ¹H NMR. The compounds underwent complete acid hydrolysis (6N HCl, 100°C, 24 h), from which the amino acid ratio was determined. The lysine ε-amino groups were then deprotected by hydrogenolysis of the protected dipeptides (I,II) (catalyst — palladium over charcoal).

The incubation 10-molar excesses of N_{α} -Cbz-lysine ethyl ester or N_{ϵ} -amino group deblocked lysyldipeptide esters with AMP, in the presence of DCC, within the reaction mixture resulted in the synthesis adenylyl- $(5'\rightarrow N_{\epsilon})$ -lysine ethyl ester (III) or adenylyl- $(5'\rightarrow N_{\epsilon})$ -lysyldipeptide esters (IV, V):

$$\begin{array}{c} O \\ | \\ HO-P-OAdo \\ OH \end{array} + \begin{array}{c} R_1 \\ | \\ NH-CH(CH_2)_4NH_2 \\ \hline DMF \\ COR_2 \end{array} \begin{array}{c} R_1 \\ | \\ NH-CH(CH_2)_4NH-P-OAdo \\ COR_2 \\ OH \end{array}$$

III. Adenylyl- $(5'\rightarrow N_s)$ - $(N_{\alpha}Cbz)$ -L-lysine ethyl ester

$$R_1 = -Cbz; R_2 = -OC_2H_5;$$

IV. Adenylyl- $(5' \rightarrow N_{\epsilon})$ - $(N_{\alpha}Boc)$ -L-lysyl-L-glutamic acid diethyl ester

$$R_1 = Boc-; R_2 = -NH-CH-C00C_2H_6;$$

 $(CH_2)_2C00C_2H_6;$

V. Adenylyl- $(5' \rightarrow N_{\epsilon})$ -(Boc)-L-threonyl-L-lysine methyl ester

$$R_1 = Boc-NH-CH-CO-; R_2 = -OCH_3;$$

 $CH(OH)CH_3$

The treatment of compounds III-V with 1N sodium hydroxide (37°C, 1 h) gave the free carboxylate analogues VI-VIII:

Compounds	Yield, ^a %	R _f ^b		T c	Ratio adenine:
		Α	В	UpAc	phosphate: amino acids
Cbz-Lys($N_{\epsilon}pA$)-OEt(III) Boc-Lys($N_{\epsilon}pA$)-Glu(OEt)-OEt(IV) Boc-Thr-Lys($N_{\epsilon}pA$)-OMe(V) Cbz-Lys($N_{\epsilon}pA$)-OH(VI) Boc-Lys($N_{\epsilon}pA$)-Glu-OH(VII) Boc-Thr-Lys($N_{\epsilon}pA$)-OH(VIII)	80 60 45 100 100	0.61 0.78 0.68 0.22 0.16 0.20	0.75 0.87 0.85 0.31 0.24 0.31	0.43 0.31 0.34 0.76 0.91 0.74	1:0.95:0.80 1:0.78:0.83:0.85 1:0.79:0.71:0.80

TABLE 1: Some characteristics of synthesized adenylyl-(5'-N_e)-lysylpepetides (III-VIII).

^aYield of compounds III-VIII was estimated spectrophotometrically after elution from chromatographic sheets. ${}^{b}R_{f}$ in solvent systems:A - 2-propanol:conc. NH₄OH:water(7:1:2); B - ethanol:1M ammonium acetate(7:3). c Electrophoretic mobility relative to adenosine-5'-monophosphate (0.05M triethylamine bicarbonate buffer pH 7.5).

VI. Adenylyl-
$$(5' \rightarrow N_{\varepsilon})$$
- $(N_{\alpha}Cbz)$ -L-lysine

$$R_1 = -Cbz; R_2 = -OH;$$

VII. Adenylyl-
$$(5' \rightarrow N_{\varepsilon})$$
- $(N_{\alpha}Boc)$ -L-lysyl-L-glutamic acid

$$R_1 = Boc-; R_2 = -NH-CH-COOH; (CH2)2COOH$$

VIII. Adenylyl-
$$(5'\rightarrow N_{\epsilon})$$
-(Boc)-L-threonyl-L-lysine

$$R_1 = Boc-NH-CH-CO-; R_2 = -OH;$$

 $CH(OH)CH_3$

Compounds III-VIII were separated from the reaction mixture by the method of paper chromatography, utilizing solvent system A (2-propanol:conc.NH₄OH:water in a volume ratio of 7:1:2). The structure of adenylyl- $(5'\rightarrow N_{\varepsilon})$ -lysine ethyl ester (III) and adenylyl- $(5'\rightarrow N_{\varepsilon})$ -lysylpeptide esters (IV, V) was determined by establishing the ratio of adenine, phosphate and amino acids after complete acid hydrolysis (6N HCl, 100°, 24 h) of these compounds. Structure of these compounds was also confirmed by ¹H NMR and by UV spectroscopic method, λ_{max} =259 nm (pH 7.0). These data along with the chromatographic and electrophoretic characteristics of synthesized compounds are presented in TABLE 1.

The hydrolytic stability of compounds III-VIII was studied. The percentage concentration of adenosine 5'-monophosphate formed as a function of acid concentration (0.01 - 0.2N HCl, 37°C, 1 h) is presented in FIGURE 1.

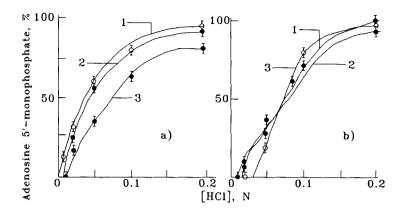


FIGURE 1. The dependance of the hydrolysis of the phosphoamide bond of compounds III-VIII upon the concentration of HCl (37°C, 1 h).

- a) 1. $Cbz-Lys(N_{\varepsilon}pA)-OEt$ (III) b) 1.
- 1. $Cbz-Lys(N_{\epsilon}pA)-OH(VI)$
 - 2. Boc-Lys(N_spA)-Glu(OEt)-(OEt) (IV)
- 2. Boc-Lys(N_spA)-Glu-OH(VII)
- 3. Boc-Thr-Lys($N_{\varepsilon}pA$)-OMe (V)
- 3. Boc-Thr-Lys($N_{\varepsilon}pA$)-OH (VIII)

Studies upon the hydrolytic characteristics of adenylyl- $(5' \rightarrow N_e)$ -lysine ethyl ester (III) and adenylyl- $(5' \rightarrow N_e)$ -lysylpeptide esters (IV, V) determined that all they undergo hydrolysis to give AMP and the respective lysine or dipeptide esters (FIGURE 1,a). Furthermore, it is of interest to note, that their respective analogues with free carboxylate groups (VI-VIII) underwent phosphoamide bound hydrolysis under these conditions to give AMP (FIGURE 1,b). These results would indicate that the lysine α -carboxylic group (VI, VIII), the threonine hydroxylic group (V) and carboxylic groups of glutamic acid (VII) are distanced from the phosphoamide center, and for this reason do not participate in the mechanism of adenylyl- $(5' \rightarrow N_e)$ -lysine (VI) and adenylyl- $(5' \rightarrow N_e)$ -lysyldipeptides (VII, VIII) hydrolysis. It is speculated that these neighbouring functional groups are located in a manner which is sterically unfavorable with respect to the phosphoramide center and can not take place in the intramolecular catalysis. In summary, the results of these experiments would then indicate that the cleavage of AMP-RNA ligase complex to form adenosine does not depend upon neighbouring threonine and glutamic acid functional groups, but rather upon other such amino acid residues, which may approach the

phosphoamide center more closely, and which are due to the tertiary structure of the RNR ligase polypeptide.

These conclusions are in agreement with those of Heaphy et al., 11 which found that replacement of glutamic acid-100 (Glu) with threonine (Thr) or glutamine (Gln) did not influence the formation of the AMP-RNA ligase complex on the first stage of the reaction.

EXPERIMENTAL

Materials. The following reactants were used in this work: adenosine-5'-phosphate (Na₂), L-glutamic acid, Boc-L-threonine, α-Cbz-L-lysine, α-Boc-ε-Cbz-L-lysine DCCA salt (Reanal, Hungary), DCC (Ferak, Berlin), TEA (Merck, West Germany), Dowex 50 (Serva, West Germany). For analytical purposes used plates "Silufol" UV 254 (Czechoslovakia) and Kieselgel 60 F₂₅₄ (Merck, West Germany). The reaction mixtures were chromatographed on paper FN 12, electrophoresis was carried out on paper FN 16 (Filtrak, GDR). In general, reactions were carried out with freshly distilled and carefully dried solvents avoiding moisture completely.

Ultraviolet spectra were recorded on a spectrophotometer DU-50 (Beckman, USA).

Nuclear magnetic resonance spectra were recorded on a Brucker 360 MHz instrument (West Germany) and chemical shifts are reported in parts per million downfield from tetramethylsilane.

The synthesis of glutamic acid diethyl ester and N_{α} -Cbz-lysine ethyl ester was carried out according to 12 .

Dipeptide Synthesis

Boc-Lys(Cbz)-Glu(OEt)-OEt(I): Boc-Lys(Cbz)-OH (1.9 g, 5 mmol), Glu(OEt)-OEt-HCl (1.2 g, 5 mmol) were dissolved in methylene chloride (100 ml). Triethylamine (0.7 ml, 5 mmol) and N,N'-dicyclohexylcarbodiimide (DCC) (1.2 g, 6 mmol) were added to the solution which had been cooled in an ice-salt bath (-10°C). The mixture was then stirred at 0-4°C overnight. The dicyclohexylurea was removed by filtration and the solvent extracted with 10% aqueous solution of sodium bisulphate (3 x 50 ml), water (3 x 50 ml) and saturated aqueous sodium bicarbonate (3 x 50 ml). The organic layer was then continuously washed with distilled water until

neutrality was obtained. The methylene chloride portion was then dried over anhydrous magnesium sulphate, after which the drying agent was removed and the solvent evaporated under reduced pressure. The resulting residue was treated with a mixture of diethyl ether and hexane, from which the precipitated product was collected, with a yield of 2.9 g (79%). ¹H NMR (DMSO) δ ppm: 8.15 (d, 1H, NH); 1.82 (m, 2H, CH_{2 β}); 2.95 (q, 2H, CH_{2 ϵ}); 1.15 (d, 3H, CH₃); 7.20 (m, 1H, NH_{ϵ}); 6.75 (d, 1H, NH_{α}); 2.38 (t, 2H, CH_{2 γ}); 1.37 (s, 9H, Boc); 4.25 (m, 1H, CH_{α}); 7.35 (m, 5H, arom.).

Boc-Thr-Lys(Cbz)-OMe (II): This peptide was synthesized by the method described above. Boc-Thr (0.654 g, 3 mmol), Lys(Cbz)-OMe•HCl (0.99 g, 3 mmol) were condensed with DCC (0.7 g, 3.5 mmol) to yield 1.1 g of the dipeptide (74%). ¹H NMR (DMSO) δ ppm: 8.74 (d, 1H, NH); 6.16 (d, 1H, CH_α); 5.01 (d, 1H, OH); 4.19 (m, 1H, CH_β); 1.12 (d, 1H, CH₃); 1.37 (s, 9H, Boc); 7.20 (m, 1H, NH_ε); 6.75 (d, 1H, NH_α); 7.35 (m, 5H, arom.); 2.95 (q, 2H, CH₂ε).

Cbz-deprotection: The protected dipeptides (I, II) (2 mmol) were dissolved in ethanol (50 ml), after which palladium over charcoal (0.1 g) added and the suspension was stirred under H_2 bubbling for 2 h at room temperature. The catalyst was filtered off and the solvent was evaporated. The lysine dipeptides with free ε -amino group were then precipitated by addition of diethyl ether-hexane mixture.

Adenylyl-(5'-N)-lysyldipeptide synthesis

Cbz-Lys(N_epA)-OEt (III): AMP (trioctylamine salt, 0.1 mmol) was dissolved in a 2 ml solution of DMF-DMSO (anhydrous, 1:1 volume ratio), to which was added N_{α}Cbz-Lys-OEt (0.3 g, 1 mmol), TEA (0.14 ml, 1 mmol), DCC (0.103 g, 0.5 mmol). The reaction mixture was incubated for 4 h at 60°C. Compound III was separated from the reaction mixture by the method of paper chromatography, utilizing solvent system A (2-propanol: conc. NH₄OH: water in a volume ratio of 7:1:2). Final yield was determined to be 80%. ¹H NMR (DMSO) δ ppm: 8.39 (s, 1H, H-8); 8.17 (s, 1H, H-2); 5.92 (d, 1H, H-1'); 3.90-4.65 (m, 5H, H-2', H-3', H-4', H-5'); 2.95 (q, 2H, CH₂₆); 1.19 (d, 3H, CH₃); 7.20 (m, 1H, NH_e); 6.75 (d, 1H, NH_{α}); 7.40 (m, 5H, arom.).

Boc-Lys(N_cpA)-Glu(OEt)-OEt (IV): Compound IV was synthesized analogously to compound III. AMP (trioctylamine salt, 0.1 mmol) was condensed with the deprotected dipeptide

- Boc-Lys-Glu(OEt)-OEt (0.56 g, 1 mmol) by reaction with DCC (0.103 g, 0.5 mmol). Compound IV was also separated from the reaction mixture by paper chromatography utilizing solvent system A, with final yield 60%. ¹H NMR (DMSO) δ ppm: 8.38 (s, 1H, H-8); 8.17 (s, 1H, H-2); 5.92 (d, 1H, H-1'); 3.90-4.60 (m, 5H, H-2', H-3', H-4', H-5'); 8.15 (d, 1H, NH); 1.82 (m, 2H, CH_{β}); 2.05 (m, 1H, CH' $_{\beta}$); 1.15 (t, 3H, CH₃); 7.20 (m, 1H, NH_{ϵ}); 6.75 (d, 1H, NH $_{\alpha}$); 2.95 (q, 2H, CH_{3ϵ}); 2.38 (t, 2H, CH_{3ϵ}); 1.37 (s, 9H, Boc).

Boc-Thr-Lys(N_epA)-OMe (V): Compound V was synthesized like compound III. AMP (trioctylamine salt, 0.1 mmol) was condensed with deprotected dipeptide - Boc-Thr-Lys-OMe (0.378 g, 1 mmol) by reaction with DCC (0.103 g, 0.5 mmol). Compound V was also separated from the reaction mixture by paper chromatography utilizing solvent system A, with the final yield of 45%. ¹H NMR (DMSO) δ ppm: 8.40 (s, 1H, H-8); 8.17 (s, 1H, H-2); 5.96 (d, 1H, H-1'), 3.98-4.71 (m, 5H, H-2', H-3', H-4', H-5'); 8.75 (d, 1H, NH); 6.16 (d, 1H, CH_{α}); 5.01 (d, 1H, OH); 4.19 (m, 1H, CH_{β}); 1.12 (d, 3H, CH₃); 1.37 (s, 9H, Boc); 7.20 (m, 1H, NH_{ϵ}); 6.75 (d, 1H, NH_{α}); 7.35 (m, 5H, arom.); 2.95 (q, 2H, CH₂ ϵ).

Synthesis of the free carboxylate analogues (VI-VIII)

Cbz-Lys(N_cpA)-OH (VI): 1 ml aqueous solution of compound V (0.02 mmol) was added to a 1 ml solution of 2N NaOH and incubated at 37°C for 1 h. The reaction mixture was then neutralized with Dowex-50 (H⁺) and compound (VI) was isolated by paper chromatography utilizing solvent system A.

Boc-Lys(N_cpA)-Glu-OH (VII) and Boc-Thr-Lys(N_cpA)-OH (VIII) were prepared analogously to compound VI.

Adenylyl- $(5' \rightarrow N_e)$ -lysylpeptide esters (III-V) and analogues with free carboxylate (VI-VIII) hydrolysis was carried out as previously described.¹³

Acknowledgments

The authors are grateful to Dr. F. Mutulis and Dr. P. Zarins for reagents and for help in peptide synthesis. We also wish to give our thanks to Dr. J. Saulytis for ¹H NMR spectra data (Institute of Organic Synthesis Academy of Sciences, Latvia).

REFERENCES

- 1. Part LIII: Juodka, B.; Bagdoniene, L. Khim, Prirod, Soed., 1990, 1, 88-93.
- 2. Silber, R.; Malathi, V. G.; Hurwitz, J. Proc. Natl. Acad. Sci. USA, 1972, 69, 3009-3013.
- 3. Uhlenbeck, O. C.; Gumport, R. J. The Enzymes, Academic Press, New York, 1982, 15, 31-60.
- 4. Venyaminova, A. G.; Yamkovoj, V. I. <u>Bioorgan. Khim.</u>, 1981, 7, 1445-1466.
- Cranston, J. W.; Silber, R.; Malathi, V. G.; Hurwitz, J. J. <u>J. Biol. Chem.</u>, 1974, 249, 7447-7456.
- Juodka, B. A.; Markuckas, A. J.; Sniechute, M. A.; Zilinskiene, V. J. <u>Bioorgan. Khim.</u>,
 1980, 6, 1733-1735.
- 7. Juodka, B. A.; Markuckas, A. J. Bioorgan. Khim., 1985, 11, 821-825.
- 8. Juodka, B. A. Nucleosides & Nucleotides, 1984, 3, 445-483.
- Thogersen, H. C.; Morris, H. R.; Rand, K. N.; Gait, M. J. <u>Eur. J. Biochem.</u>, 1985, <u>147</u>, 325-329.
- 10. Rand, K. N.; Singh, M.; Thogersen, H. C.; Gait, M. J. Proc. Int. Symp. Biomol. Struct.

 Interactions. Suppl. J. Biosci., 1985, 8, 89-106.
- 11. Heaphy, S.; Singh, M. Gait, M. J. Biochemistry, 1987, 26, 1688-1696.
- 12. Greenstein, J. P.; Winitz, M. "Chemistry of the Amino Acids", 1961, eds. John Wiley and Sons, Inc., 424-427.
- Juodka, B. A.; Sasnauskiene, S. J. <u>J. Carbohydr. Nucleosides, Nucleotides</u>, 1981, 8, 19-39.

Received 5/20/90 Accepted 6/17/92